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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
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10/771,544

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Maki Hoshino

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FOLEY AND LARDNER LLP
SUITE 500
3000 K STREET NW
WASHINGTON, DC 20007

EXAMINER

HENDRICKSON, STUART L

ART UNIT

PAPER NUMBER

1793

MAIL DATE

DELIVERY MODE

12/04/2007

PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Office Action Summary	Application No. 10/771,544	Applicant(s) HOSHINO, MAKI	
	Examiner Stuart Hendrickson	Art Unit 1793	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 22 October 2007.
- 2a) ☐ This action is **FINAL**. 2b) ☒ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1-4, 6 and 8-12 is/are pending in the application.
- 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
- 5) ☐ Claim(s) _____ is/are allowed.
- 6) ☒ Claim(s) 1-4, 6, 8-12 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on _____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All b) ☐ Some * c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
 2. ☐ Certified copies of the priority documents have been received in Application No. _____.
 3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

- | | |
|--|---|
| 1) <input type="checkbox"/> Notice of References Cited (PTO-892) | 4) <input type="checkbox"/> Interview Summary (PTO-413)
Paper No(s)/Mail Date. _____ |
| 2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948) | 5) <input type="checkbox"/> Notice of Informal Patent Application |
| 3) <input type="checkbox"/> Information Disclosure Statement(s) (PTO/SB/08)
Paper No(s)/Mail Date _____ | 6) <input type="checkbox"/> Other: _____ |

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The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action. The RCE is accepted.

Claims 1-4 and 8-12 are rejected under 35 U.S.C. 103(a) as being unpatentable over Takamura 6,548,034 in view of Arntz 5,364,984.

Takamura '034 teaches a method of reducing the CO concentration of a mixed gas containing hydrogen, CO and oxygen with a Pt catalyst (see col. 4, lines 4-13) and the gas is supplied at 100 – 100,000 [1/h] (lines 59-62) at a temperature of 40°C to 200°C. Takamura does not discuss the CO adsorption amount. It appears possessed since some examples use 1% Pt, which is exactly the amount used by most of the examples of the present specification. Also, Arntz '984 teaches Pt catalyst with CO adsorption of 0.5 to 1.6 ml of CO/g of catalyst (see col. 3 lines 50-55, col. 6 lines 5-9, and col. 10 lines 3-7). It can be seen from the amount of Pt and the capacity of Arntz that the catalyst of Takamura is expected to meet this limitation. As to claim 2, Takamura teaches a CO concentration of about 1 mol% (col. 5, lines 33-36). The oxygen is 0.5 to 4 times the CO concentration (col. 4, lines 55-57). Also taught are the additional metals.

As to claim 8, in Takamura '034, the carbon monoxide reducing catalyst is a monolithic catalyst (see ex. 2 in col. 8, and col. 4, lines 4-37 showing a supported catalyst); As to claim 8, if the amount is different- then it is an obvious expedient to use the claimed amount to optimize performance with expense. The amount of the second component appears to fall within the range of 2g or less per liter of the monolithic catalyst (see col. 4, lines 45-49, as compared to example 1 in the present disclosure). The source of claims 9 and 10 does not in and of itself distinguish the actual gas composition; treating a CO containing gas from any source is an obvious expedient, as long as the composition itself is similar.

No difference of any consequence is seen between claim 1 and 12. Takamura differs in not specifically teaching the claimed numerical parameters in one example, however teaches ranges for all which overlap the claimed ranges. Thus, the claims are unpatentable- In re Malagari 182 USPQ 549; it would have been obvious to one of ordinary skill in the art at the time the invention was made to vary the amounts catalyst so as to arrive at the claimed ratios

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and material, since it has been held that discovering an optimum value or a result effective variable involved only routine skill in the art. See, e.g., In re Boesch, 617 F.2d 272, 205 U.S.P.Q. 215 (CCPA 1980). The artisan would have been motivated to vary the amount of the second component of the catalyst because varying it would optimize the amount of carbon monoxide adsorbed, while reducing the cost of the catalyst.

Claim 6 is rejected under 35 U.S.C. 103(a) as being unpatentable over Takamura taken with Arntz as applied to claims 1-4, 8-12 above, and further in view of USP 6,913,739 to Shore. As to claim 6, Takamura '034 does not teach that the second component is a rare earth. Shore '739 teaches the use of a rare earth metal as a catalyst (see column 3, lines 57-61). Shore '739 further discloses that the rare earth metal can be lanthanum, cerium, neodymium, and praseodymium (see column 5, line 64 – column 6, line 5). As cerium (and other rare earth metals) are known to be effective, yet less expensive preferential oxidation catalysts (see column 3, lines 11-16), it would have been obvious to one of ordinary skill in the art to combine the rare earth metal catalyst component of Shore '739 with the catalyst composition of Takamura '034 in order to reduce the carbon monoxide composition in a gas stream effectively with reduced cost. It is further noted that when A is known (transition metal) and B (rare earth metal) is known to perform a function (carbon monoxide oxidation), then A and B together is obvious. See, e.g., In re Kerkhoven, 205 U.S.P.Q. 1069 (CCPA 1980).

Claim 10 is rejected under 35 U.S.C. 103(a) as being unpatentable over Takamura taken with Arntz as applied to claims 1-4, 8-12 above, and further in view of USP 3,584,608 to Shibagaki.

As to claim 10, Takamura '034 teaches that the CO containing gas can be supplied by a process for reforming hydrogen methanol, et cetera (see column 1, lines 11-15); therefore, Takamura '034 suggests that it does not matter where the detrimental CO containing gas stream comes from. However, Takamura does not explicitly disclose reducing the CO concentration from the exhaust of an internal combustion engine. Shibagaki discloses removing

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detrimental components such as CO within the exhaust of an internal combustion engine (see col. 1 lines 35 – 51). As such, it would have been obvious to one of ordinary skill in the art at the time of this invention to use the Takamura process to reduce the CO concentration in a carbon monoxide containing gas stream from an internal combustion engine, as in Shibagaki, in order to remove these detrimental components as desired in both Takamura and Shibagaki.

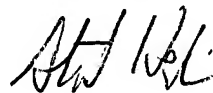
Claims 1-4, 6, 8-12 are rejected under 35 U.S.C. 103(a) as being unpatentable over Shore 6913739 in view of Arntz 5364984.

Shore teaches, especially in col. 25-26, oxidizing CO in the claimed atmosphere. Also taught are overlapping temperature, velocity and amounts of metals, which render the claims upatentable; In re Malagari 182 USPQ 549. The CO adsorption amount appears possessed, as evidenced by Arntz as explained above.

Claim 3 is objected to under 37 CFR 1.75(c), as being of improper dependent form for failing to further limit the subject matter of a previous claim. Applicant is required to cancel the claim(s), or amend the claim(s) to place the claim(s) in proper dependent form, or rewrite the claim(s) in independent form. The claim expands the lower limit from 0.1 to 0.05.

Applicant's arguments filed 10/22/07 have been fully considered but they are not persuasive. The new rejections reflect the amendments to the claims, which are now obvious rather than anticipated. The references teach 5% of the additional metal. The arguments overlook these teachings and focus only on the specific examples.

Any inquiry concerning this communication should be directed to examiner Hendrickson at telephone number (571) 272-1351.



Stuart Hendrickson
examiner Art Unit 1793